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## Probing Excitation Delocalization in Supramolecular Chiral Stacks by Means of Circularly Polarized Light: Experiment and Modeling [*J. Am. Chem. Soc.* 2007, *129*, 7044–7054].

Frank C. Spano, Stefan C. J. Meskers, Emanuelle Hennebicq, and David Beljonne

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Probing Excitation Delocalization in Supramolecular Chiral Stacks by Means of Circularly Polarized Light: Experiment and Modeling [*J. Am. Chem. Soc.* 2007, *129*, 7044–7054]. Frank C. Spano,\* Stefan C. J. Meskers, Emanuelle Hennebicq, David Beljonne

A sign error in the fortran code used to evaluate  $g_{abs}(\omega)$  and  $g_{lum}(\omega)$  has been discovered. Since  $g_{abs}(\omega)$  and  $g_{lum}(\omega)$  change sign when the helical pitch angle  $\phi$  changes sign, the values for  $\phi$  reported for all calculations need to be negated. With these corrected values of  $\phi$  all figures remain unchanged. We therefore obtain good agreement with the measured  $g_{abs}(\omega)$  and  $g_{lum}(\omega)$ -for a pitch angle of  $+14^{\circ}$  (left-handed helix) and not  $-14^{\circ}$  (right-handed helix) as originally reported.

Additional support for the left-handed helix was obtained from molecular dynamics calculations. We started with a 12molecule MOPV4 stack built from enantiomerically pure MOPV4 molecules bearing exclusively (S)-2-methylbutoxy side chains. Two structures were generated with intermolecular angles ( $\phi$ ) along the stack direction of either 20° or -20°. A molecular dynamics simulation of 50 ps at 300 K (within the NVT ensemble, Dreiding force field,<sup>1</sup> Gasteiger charges<sup>2</sup>) was then performed for both the left-handed and right-handed helices. The most stable structures generated from the MD run were subsequently optimized at 0 K using the same force field. In the optimized structures, the rotation angle is reduced to an average value of  $\sim 15^{\circ}$  (left) and  $\sim -13^{\circ}$  (right). The calculations indicate that the left-handed helix is more stable by  $\sim$ 3 kcal/ mol per molecule than its right-handed counterpart and that the energy difference arises primarily from less favorable van der Waals interactions.

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Synthesis of Monoalkoxide Monopyrrolyl Complexes Mo(NR)(CHR')(OR")(pyrrolyl): Enyne Metathesis with High Oxidation State Catalysts [*J. Am. Chem. Soc.* 2007, *129*, 12654–12655] Rojendra Singh, Richard R. Schrock,\* Peter Müller, and Amir H. Hoveyda

Page 126555. In ref 4 "Blanc, R." should be "Blanc, F."

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